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section 4.3 how atoms differ in your textbook, read about atomic number. ... use the figures to answer the following questions. osmium 76 os 190.2 niobium 41 nb 92.906

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sodium atoms ...

PROTONS, NEUTRONS, AND ELECTRONS—HOW ATOMS DIFFER

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... answer each question on the lines provided. ... protons, neutrons, and electrons—how atoms differ ... and electrons—how atoms differ

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the structure of the atom the structure of the atom ... different atoms ... explain your answer.

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section 3: how atoms differ how atoms differ. 1. what is the name for atoms of an element that have different masses? need a hint? a) isotopes: b) isomers: c ...

CH 9, LESSON 2 - PROTONS, NEUTRONS, AND ELECTRONS - HOW ...

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... lesson 2 - protons, neutrons, and electrons - how atoms differ. ... neutrons, and electrons - how atoms differ. study. play. ... protons, neutrons, and electrons ...

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... how atoms differ (chapter 4, section 3) atomic number •elements are in order by their atomic number on the periodic table. atomic number (z) = number of ...

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section 4.3 how atoms differ in your textbook, read about atomic number. for each statement below, write true or

false. 1. ... answer the following questions. 6.

CH 4-3 ELEMENTS, ISOTOPES, AND IONS-HOW ATOMS DIFFER ...

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ch 4-3 elements, isotopes, and ions-how atoms differ. study. play. element. ... a substance made up of atoms of two or more different elements joined by chemical bonds.

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